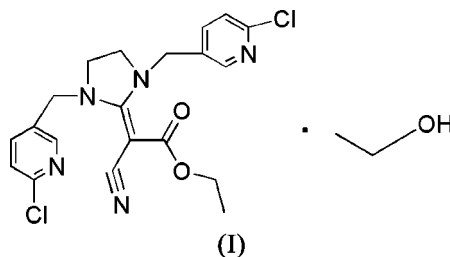


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## Key indicators

Single-crystal X-ray study  
 $T = 292$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
Disorder in solvent or counterion  
 $R$  factor = 0.055  
 $wR$  factor = 0.167  
Data-to-parameter ratio = 17.2For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Ethyl 2-{1,3-bis[(6-chloro-3-pyridyl)methyl]-  
imidazolidin-2-ylidene}-2-cyanoacetate  
ethanol solvateIn the title compound,  $\text{C}_{20}\text{H}_{19}\text{Cl}_2\text{N}_5\text{O}_2 \cdot \text{C}_2\text{H}_5\text{OH}$ , all atoms of  
the ethanol solvent molecule are disordered over two  
positions. Intermolecular  $\text{O}-\text{H} \cdots \text{O}$  and  $\text{C}-\text{H} \cdots \text{O}$  hydrogen  
bonds and  $\text{O}-\text{H} \cdots \pi$  interactions contribute to the stability of  
the crystal structure.Received 3 April 2006  
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## Comment

Imidacloprid is an example of a highly effective low-toxicity  
neonicotinoid insecticide, marketed by the Bayer company  
since 1984 (Tang, 2002; Chao *et al.*, 2002). Subsequently, many  
neonicotinoid analogues have been developed (Shiokawa *et al.*,  
1986). The 3-(aminomethyl)pyridine group proved to be an  
active moiety in these insecticides (Yamamoto *et al.*, 1994),  
with the two N atoms of the active group interacting with the  
nicotine acetylcholine receptor (nAChR) of the insects with  
fatal effects. Here, we report the crystal structure of a related  
chloromethylpyridine derivative, the title compound, (I),  
which was synthesized by the reaction of ethyl 2-cyano-2-  
(imidazolidin-2-ylidene)acetate with 2-chloro-5-(chloro-  
methyl)pyridine.Atoms C7, C8, N2, N3, C15 of the five-membered imida-  
zolidine ring of (I) are almost coplanar, with an r.m.s. devia-  
tion of 0.0180 Å. The dihedral angles between the  
imidazolidine and N1- and N4-pyridine rings are 113.42 (14)  
and 114.01 (15)°, respectively.In the crystal structure,  $\text{O3}-\text{H3A} \cdots \text{O1}$  hydrogen bonds  
between the ethanol solvent molecule and the acetate group  
and  $\text{C19}-\text{H19B} \cdots \text{O1}$  hydrogen bonds between adjacent  
molecules of (I) stabilize the structure (Table 1). Furthermore,  
there is evidence for an  $\text{O}-\text{H} \cdots \pi$  interaction involving the  
minor component of the disordered ethanol solvate molecule,  
with  $\text{O3}'-\text{H3}' \cdots \text{Cg1} = 3.392$  (9) Å, where Cg1 is the centroid  
of the N2/C7/C8/N3/C15 ring.

## Experimental

A solution of 2-chloro-5-chloromethylpyridine (10 mmol) in anhy-  
drous dimethylformamide (10 ml) was added dropwise to a mixture

of ethyl 2-cyano-2-(imidazolidin-2-ylidene)acetate (5 mmol) and sodium hydride (10 mmol) in anhydrous dimethylformamide (20 ml) in an ice-bath. The mixture was then stirred at room temperature until thin-layer chromatography indicated that the reaction was complete. The solvent was removed, water added and the product mixture extracted into chloroform. The product was purified by flash column chromatography on silica gel using petroleum ether–ethyl acetate (1:2 v/v) as eluent, to give a white solid (yield 62%, m.p. 429 K). Crystals of compound (I) were obtained from a solution in absolute ethanol.

#### Crystal data

$C_{20}H_{19}Cl_2N_5O_2 \cdot C_2H_6O$	$Z = 4$
$M_r = 478.37$	$D_x = 1.369 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 11.6424 (10) \text{ \AA}$	$\mu = 0.31 \text{ mm}^{-1}$
$b = 18.1817 (15) \text{ \AA}$	$T = 292 (2) \text{ K}$
$c = 11.9659 (10) \text{ \AA}$	Block, colourless
$\beta = 113.573 (1)^\circ$	$0.30 \times 0.30 \times 0.20 \text{ mm}$
$V = 2321.6 (3) \text{ \AA}^3$	

#### Data collection

Bruker SMART CCD area-detector diffractometer	5526 independent reflections
$\varphi$ and $\omega$ scans	4039 reflections with $I > 2\sigma(I)$
Absorption correction: none	$R_{\text{int}} = 0.060$
26922 measured reflections	$\theta_{\text{max}} = 28.0^\circ$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.098P)^2 + 0.0282P]$
$R[F^2 > 2\sigma(F^2)] = 0.055$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.167$	$(\Delta/\sigma)_{\text{max}} = 0.003$
$S = 1.10$	$\Delta\rho_{\text{max}} = 0.47 \text{ e \AA}^{-3}$
5526 reflections	$\Delta\rho_{\text{min}} = -0.47 \text{ e \AA}^{-3}$
321 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	(Sheldrick, 1997)
	Extinction coefficient: 0.0100 (15)

**Table 1**

Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ ).

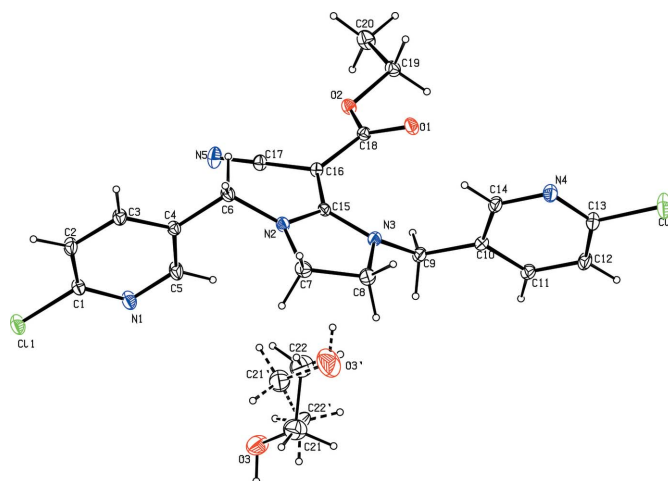
$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C9-H9A \cdots O1$	0.97	2.46	2.944 (2)	111
$C19-H19B \cdots O1^i$	0.97	2.57	3.448 (2)	150
$O3-H3A \cdots O1^{ii}$	0.82	2.30	3.086 (4)	160

Symmetry codes: (i)  $-x + 1, -y, -z + 1$ ; (ii)  $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ .

All H atoms bound to C and O were refined using a riding model, with  $C-H = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for aromatic H atoms,  $C-H = 0.97 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for  $\text{CH}_2$  and  $C-H = 0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for  $\text{CH}_3$  atoms, and with  $O-H = 0.82 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$  for the OH groups. All atoms in the ethanol molecule are disordered over two positions. The occupancy factors refined to 0.613 (5)/0.387 (5).

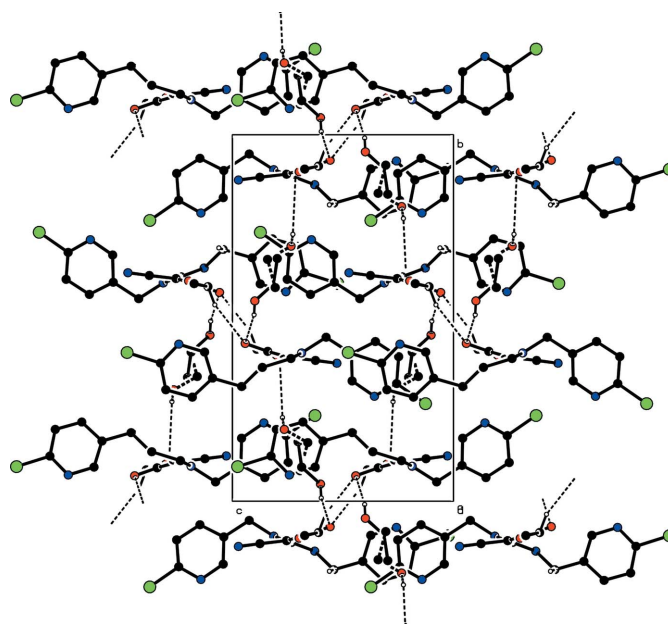
Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT-Plus* (Bruker, 2001; data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PLATON*.

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**Figure 1**

The structure of (I), showing the atom-labelling scheme and with displacement ellipsoids drawn at the 30% probability level. The minor disorder component of the ethanol solvent molecule is shown with bonds represented by dashed lines.



**Figure 2**

A packing diagram for (I). Hydrogen bonds and  $O-H \cdots \pi$  interactions are shown as dashed lines. H atoms not involved in these interactions have been omitted.

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